Acta Crystallographica Section E

Structure Reports Online

ISSN 1600-5368

N-(4-Chlorophenyl)-2,2,2-trimethylacetamide

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Kev indicators

Single-crystal X-ray study T = 299 KMean $\sigma(C-C) = 0.003 \text{ Å}$ R factor = 0.042wR factor = 0.133 Data-to-parameter ratio = 14.4

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

The structure of the title compound, $C_{11}H_{14}ClNO$, is closely related to the ring-unsubstituted 2,2,2-trimethyl-N-phenylacetamide and 2,2,2-trimethyl-N-(4-methylphenyl)acetamide with slightly different bond parameters. The molecules are linked into chains running along the b-axis direction through N-H···O hydrogen bonding.

Received 23 March 2007 Accepted 29 March 2007

Comment

In the present work, the structure of N-(4-chlorophenyl)-2,2,2trimethylacetamide (4CPTMA), (I), has been determined as part of a study to systematize the structures of N-aromatic amides (Gowda et al., 2004, 2006; Gowda, Kozisek, Svoboda & Fuess, 2007). The structure of 4CPTMA (Fig. 1) is closely related to the ring-unsubstituted 2,2,2-trimethyl-N-phenylacetamide (Gowda, Paulus et al., 2007) and 2,2,2-trimethyl-N-(4-methylphenyl)acetamide (Gowda, Kozisek, Tokarcik & Fuess, 2007) with similar geometric parameters. In 4CPTMA, the molecules are linked into chains along the b axis through N-H···O hydrogen bonds

$$CI \longrightarrow \begin{matrix} H \\ N \\ -C \\ C(CH_3) \end{matrix}$$

Experimental

The title compound was prepared according to the literature method (Gowda et al., 2003). The purity of the compound was checked by determining its melting point. It was characterized by recording its IR and NMR spectra (Gowda et al., 2003). Single crystals of the title compound were obtained by slow evaporation of an ethanol solution and used for X-ray diffraction analysis at room temperature.

Crystal data

C₁₁H₁₄ClNO $V = 2308.2 (3) \text{ Å}^3$ $M_{\rm m} = 211.68$ Z = 8Orthorhombic, Pbca Cu Kα radiation a = 9.6933 (6) Å $\mu = 2.67 \text{ mm}^$ b = 10.115 (1) ÅT = 299 (2) Kc = 23.542 (1) Å $0.77 \times 0.25 \times 0.17 \text{ mm}$

Data collection

Nonius CAD-4 diffractometer 1730 reflections with $I > 2\sigma(I)$ $R_{\rm int} = 0.016$ Absorption correction: ψ scan (North et al., 1968) 3 standard reflections $T_{\min} = 0.555, T_{\max} = 0.661$ frequency: 120 min 2104 measured reflections intensity decay: 1.0% 2061 independent reflections

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Acta Cryst. (2007). E63, o2329-o2330

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Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.042$ $wR(F^2) = 0.133$ S = 1.082061 reflections 143 parameters H atoms treated by a mixture of independent and constrained refinement

 $\Delta \rho_{\text{max}} = 0.25 \text{ e Å}^{-3}$ $\Delta \rho_{\text{min}} = -0.32 \text{ e Å}^{-3}$

Table 1 Hydrogen-bond geometry (Å, °).

$D-\mathrm{H}\cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D-\mathrm{H}\cdots A$
N7−H7 <i>N</i> ···O6 ⁱ	0.84 (2)	2.14 (2)	2.9773 (19)	172 (2)

Symmetry code: (i) $-x + \frac{3}{2}$, $y + \frac{1}{2}$, z.

All C-bound H atoms were positioned geometrically and treated as riding, with C-H = 0.93 (CH aromatic) or 0.96 Å (CH₃) and $U_{\rm iso}({\rm H}) = 1.2 U_{\rm eq}({\rm C})$ or $1.5 U_{\rm eq}({\rm methyl~C})$. The H atom of the NH group was located in a difference map and its position refined [N-H = 0.84 (2) Å and $U_{\rm iso}({\rm H}) = 1.2 U_{\rm eq}({\rm N})$].

Data collection: *CAD-4-PC* (Nonius, 1996); cell refinement: *CAD-4-PC*; data reduction: *REDU4* (Stoe & Cie, 1987); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *SHELXL97*.

BTG gratefully thanks the Alexander von Humboldt Foundation, Bonn, Germany, for a research fellowship.

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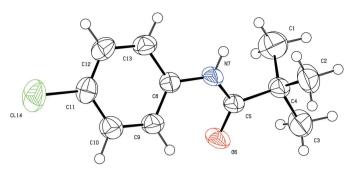


Figure 1

The molecular structure of the title compound, showing the atomlabelling scheme. Displacement ellipsoids are drawn at the 30% probability level and H atoms are represented as small spheres of arbitrary radius.

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